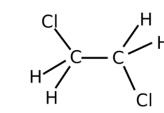


Ethylene Dichloride (EDC), also known as 1,2 dichloroethane, is a chlorinated hydrocarbon. Under normal temperature and pressure conditions, it is a colourless liquid.

Key Properties

| Chemical formula | |
|------------------|--|
| CAS number | |
| Molar mass | |
| Density | |
| Melting point | |
| Boiling point | |
| | |



Product Safety

EDC is a highly flammable, toxic and possibly carcinogenic product. It causes skin, eye and respiratory irritation. For more details and information on precautions to be taken, please refer to our Material Safety Data Sheet which is available on request.

Production

EDC is produced by chlorination of ethylene, which can be done via two routes:

 $C_2H_4Cl_2$

-35°C 84°C

107-06-2

98.95 g/mol

1253 kg/m³

- Direct chlorination, using chlorine gas (Cl₂)
- Oxychlorination, using hydrochloric acid (HCl)

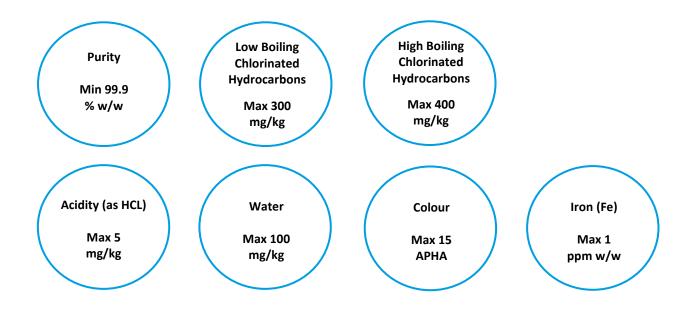
Applications

The main application for EDC is the production of vinyl chloride, often referred to as Vinyl Chloride Monomer (VCM), which is the raw material for producing PVC.

It is also used as an intermediate in the production of organic derivatives.

Quality

The table below shows the most important quality parameters of the Vynova EDC specification. A more detailed specification can be obtained on request.



Contact us :

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